

Mechanistic studies of conformationally controlled ionic cycloadditions and state selective reactions

A. Kilaj¹, H. Gao¹, D. Rösch¹, U. Rivero¹, J. Küpper^{2,3*}, S. Willitsch^{1*}

¹Department of Chemistry, University of Basel, Klingelbergstrasse 80, 4056 Basel, Switzerland, ²Center for Free-Electron Laser Science, DESY, Notkestrasse 85, 22607 Hamburg, Germany, ³Department of Physics, University of Hamburg, Luruper Chaussee 149, 22761 Hamburg, Germany

Despite their significance in organic synthesis, the mechanistic details of Diels-Alder cycloadditions, in which a diene and a dienophile react to form a cyclic product, still remain an unsolved and extensively discussed question. It has proven difficult to determine whether only the *s-cis* conformer (concerted mechanism) or both conformers (stepwise mechanism) of the involved diene react to form the cyclic product. Due to their high stereo- and regioselectivity, a concerted mechanism is generally assumed [1]. For polar Diels-Alder reactions, however, where one of the reactants contains a charge, the reaction may instead favor a stepwise mechanism [2].

In order to shed light on these questions we investigate the reactivities of individual conformers by employing molecular beam technology [3]. Owing to their different dipole moments, conformers can be spatially separated in an inhomogeneous electric field. Directing a molecular beam of spatially separated conformers into a cloud of sympathetically cooled molecular ions in an ion trap allows us to study highly conformationally selective reactions such as Diels-Alder cycloadditions. Previously, this method has been successfully applied to study reactive collisions of 3-aminophenol with a Coulomb crystal of Ca⁺ ions [4][5]. As a representative example, we want to study the conformationally resolved ionic cycloaddition reaction of 2,3-dibromo-1,3-butadiene which exhibits two different conformers, *s-cis* and *s-trans*, with ionic maleic anhydride. The products of the reactive collisions are analyzed by an integrated high-resolution time-of-flight mass-spectrometer. The performance of the setup was recently confirmed by measuring the rate constant of the reaction between laser cooled Ca⁺ ions with N₂O [6]. This advancement will enable us to trace the complex mechanistic pathways of Diels-Alder reactions.

Besides conformers, our setup also enables us to study state selective reactions. In the presence of a strong inhomogeneous electric field, water molecules in a molecular beam can be separated into their two nuclear spin isomers *ortho* and *para* which correlate with different rotational states [7]. Currently we are investigating the reaction of rotational state selected H₂O molecules with N₂⁺ and N₂H⁺ ions. The flexibility of our setup allows us to access the individual spin isomers and study their reactivities.

[1] M. Eberlin, *Int. Journal of Mass Spectrometry*, **2004**, 235, 263.

[2] U. Rivero et al., *Chem. Phys. Lett.*, **2017**, dx.doi.org/10.1016/j.cplett.2017.03.063.

[3] F. Filsinger et al., *Phys. Rev. Lett.*, **2008**, 100, 133003.

[4] Y.-P. Chang et al., *Science*, **2013**, 342, 98.

[5] D. Rösch et al., *JCP*, **2014**, 140, 124202).

[6] D. Rösch et al., *EPJ techniques and instrumentation*, **2016**, 3:5.

[7] D. A. Horke et al., *Angew. Chem. Int. Ed.*, **2014**, 53, 11965- 11968.